

# Atomic Structure Puzzle

<p>Atomic number for Cl</p> <p>_____</p> <p>Set 1</p>	<p>Mass number for <math>^{16}_8\text{O}</math></p> <p>_____</p> <p>Set 1</p>	<p>Number of neutrons in <math>^{46}_{20}\text{Ca}</math></p> <p>_____</p> <p>Set 1</p>	<p>Number of electrons in a neutral cerium atom</p> <p>_____</p> <p>Set 1</p>
<p>Number of protons in a <math>\text{Ba}^{2+}</math> ion</p> <p>_____</p> <p>Set 1</p>	<p>Atomic number of a +1 ion having 18 electrons</p> <p>_____</p> <p>Set 1</p>	<p>Mass number of an atom with 1 proton, 1 neutron, and 1 electron</p> <p>_____</p> <p>Set 1</p>	<p>Number of protons in an element with atomic number 12</p> <p>_____</p> <p>Set 1</p>
<p>Atomic number of a +3 ion with 30 neutrons and 27 electrons</p> <p>_____</p> <p>Set 1</p>	<p>Number of neutrons in <math>^{75}\text{As}</math></p> <p>_____</p> <p>Set 1</p>	<p>Mass number of a neutral atom with 30 neutrons and 25 electrons</p> <p>_____</p> <p>Set 1</p>	<p>Number of electrons in a <math>\text{O}^{2-}</math> ion</p> <p>_____</p> <p>Set 1</p>
<p>Number of electrons in a neutral nickel atom</p> <p>_____</p> <p>Set 1</p>	<p>Number of protons in a neutral atom with 82 neutrons and 57 electrons</p> <p>_____</p> <p>Set 1</p>	<p>Number of protons in <math>\text{Co}^{3+}</math></p> <p>_____</p> <p>Set 1</p>	<p>Atomic number of <math>^{36}_{18}\text{Ar}</math></p> <p>_____</p> <p>Set 1</p>
<p>Number of protons in <math>\text{Cu}^+</math></p> <p>_____</p> <p>Set 1</p>	<p>Mass number of an ion with 28 protons, 32 neutrons, and 26 electrons</p> <p>_____</p> <p>Set 1</p>	<p>Number of neutrons in <math>^{64}\text{Zn}^{2+}</math></p> <p>_____</p> <p>Set 1</p>	<p>Number of electrons in a neutral atom with atomic number 52</p> <p>_____</p> <p>Set 1</p>

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

Atomic number of potassium  _____ Set 2	Number of protons in $^{23}\text{Na}^+$  _____ Set 2	Mass number of an atom with 4 protons, 4 electrons and 5 neutrons  _____ Set 2	Number of neutrons in $^{15}_7\text{N}$  _____ Set 2
Number of electrons in $\text{F}^-$  _____ Set 2	Number of electrons in a neutral magnesium atom  _____ Set 2	Number of electrons in $\text{P}^{3-}$  _____ Set 2	Mass number of an ion with 8 protons, 10 electrons, and 9 neutrons  _____ Set 2
Number of protons in a neutral atom with 30 neutrons and 25 electrons  _____ Set 2	Number of neutrons in $^{37}\text{Cl}$  _____ Set 2	Number of protons in an element with atomic number 13  _____ Set 2	Number of electrons in a neutral iron atom  _____ Set 2
Number of neutrons in a $^{41}\text{K}^+$ ion  _____ Set 2	Atomic number of a $-2$ ion having 18 electrons  _____ Set 2	Number of protons in a $\text{Sc}^{3+}$ ion  _____ Set 2	Atomic number of $^{59}_{27}\text{Co}$  _____ Set 2
Mass number of $^{14}_7\text{N}$  _____ Set 2	Mass number of a $+1$ ion with 12 neutrons, and 10 electrons  _____ Set 2	Mass number of a neutral atom with 12 neutrons and 12 electrons  _____ Set 2	Number of electrons in a neutral atom with atomic number 15  _____ Set 2

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

Number of protons in a $K^+$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of electrons in a neutral atom with atomic number 21  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of electrons in a neutral carbon atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of protons in an element with atomic number 25  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>
Number of protons in a silicon atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of electrons in a $Mg^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of neutrons in a $^{17}O^{2-}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Atomic number of a $-2$ ion with 18 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>
Mass number of a neutral atom with 12 neutrons and 10 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of electrons in a neutral vanadium atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of neutrons in $^{33}S$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of protons in a neutral atom with 24 neutrons and 20 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>
Mass number of a $+2$ ion with 14 neutrons and 10 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Atomic number of zinc  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Mass number of $^{56}_{26}Fe$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of neutrons in $^{29}_{14}Si$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>
Mass number of an atom with 6 protons, 7 neutrons, and 6 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Atomic number of $^{53}_{24}Cr$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Number of electrons for $^{37}Cl^-$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>	Mass number of an ion with 6 protons, 6 electrons, and 6 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 3</span>

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

<p>Atomic number of silver</p> <p>_____</p> <p>Set 4</p>	<p>Number of neutrons in <math>^{91}\text{Zr}</math></p> <p>_____</p> <p>Set 4</p>	<p>Number of protons in <math>\text{Cl}^-</math></p> <p>_____</p> <p>Set 4</p>	<p>Mass number of a neutral atom with 36 neutrons and 30 electrons</p> <p>_____</p> <p>Set 4</p>
<p>Number of electrons in a neutral terbium atom</p> <p>_____</p> <p>Set 4</p>	<p>Number of neutrons in <math>^{88}_{38}\text{Sr}</math></p> <p>_____</p> <p>Set 4</p>	<p>Mass number of an ion with 27 protons, 25 electrons, and 32 neutrons</p> <p>_____</p> <p>Set 4</p>	<p>Mass number of <math>^{60}_{28}\text{Ni}</math></p> <p>_____</p> <p>Set 4</p>
<p>Mass number of an atom with 24 protons, 30 neutrons and 24 electrons</p> <p>_____</p> <p>Set 4</p>	<p>Number of protons in a neutral atom with 80 neutrons and 56 electrons</p> <p>_____</p> <p>Set 4</p>	<p>Number of electrons in a neutral indium atom</p> <p>_____</p> <p>Set 4</p>	<p>Number of protons in a <math>\text{Rh}^{3+}</math> ion</p> <p>_____</p> <p>Set 4</p>
<p>Number of electrons in a <math>\text{Ag}^+</math> ion</p> <p>_____</p> <p>Set 4</p>	<p>Atomic number of <math>^{165}_{67}\text{Ho}</math></p> <p>_____</p> <p>Set 4</p>	<p>Number of electrons in a +2 ion of tin</p> <p>_____</p> <p>Set 4</p>	<p>Number of electrons in a neutral atom with atomic number 58</p> <p>_____</p> <p>Set 4</p>
<p>Mass number of a +4 ion with 72 neutrons and 48 electrons</p> <p>_____</p> <p>Set 4</p>	<p>Mass number of an atom with 32 protons and 42 neutrons</p> <p>_____</p> <p>Set 4</p>	<p>Atomic number of a -1 ion having 54 electrons</p> <p>_____</p> <p>Set 4</p>	<p>Number of protons in a tellurium atom</p> <p>_____</p> <p>Set 4</p>

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

Number of electrons in $\text{Ba}^{2+}$  _____ Set 5	Number of neutrons in $^{143}_{60}\text{Nd}$  _____ Set 5	Mass number of an ion with 30 protons, 28 electrons, and 38 neutrons  _____ Set 5	Number of protons in a neutral atom with 111 neutrons and 76 electrons  _____ Set 5
Mass number of a $-1$ ion with 44 neutrons and 36 electrons  _____ Set 5	Mass number of an atom with 37 protons, 37 electrons, and 49 neutrons  _____ Set 5	Number of electrons in a neutral europium atom  _____ Set 5	Mass number of $^{70}_{32}\text{Ge}$  _____ Set 5
Number of neutrons in $^{125}\text{Te}$  _____ Set 5	Number of neutrons in $^{133}_{55}\text{Cs}^+$  _____ Set 5	Number of protons in $\text{Pb}^{4+}$  _____ Set 5	Atomic number of $^{175}_{71}\text{Lu}$  _____ Set 5
Number of electrons in a neutral iridium atom  _____ Set 5	Mass number of a neutral atom with 49 neutrons and 38 electrons  _____ Set 5	Atomic number of thallium  _____ Set 5	Number of protons in an atom with atomic number 72  _____ Set 5
Atomic number of a $+4$ ion with 70 electrons  _____ Set 5	Number of electrons in a neutral At atom  _____ Set 5	Number of electrons in a neutral atom with atomic number 75  _____ Set 5	Number of protons in a $\text{Hg}^+$ ion  _____ Set 5

# Atomic Structure Puzzle

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**a i r c**  
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**c n n o**

# Atomic Structure Puzzle

Atomic number of silver  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of electrons in a neutral atom with atomic number 58  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of neutrons in $^{97}_{42}\text{Mo}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Atomic number of $^{144}_{60}\text{Nd}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>
Number of electrons in a neutral rhodium atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of neutrons in $^{86}_{37}\text{Rb}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Atomic number of a $-1$ ion having 54 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Mass number of a neutral atom with 38 neutrons and 31 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>
Mass number of an atom with 22 protons, 22 electrons, and 24 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of protons in an element with atomic number 59  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of electrons in a $\text{Cs}^+$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of protons in $\text{Sn}^{4+}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>
Number of electrons in a neutral zirconium atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of electrons in $\text{Sn}^{2+}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of protons in an $\text{Sb}^{3+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Mass number of an ion with 26 protons, 23 electrons and 31 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>
Number of neutrons in a $^{109}\text{Ag}^+$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Mass number of $^{63}_{29}\text{Cu}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Number of protons in a neutral atom with 68 neutrons and 52 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>	Mass number of a $\text{Ce}^{4+}$ ion with 82 neutrons and 54 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 6</span>

# Atomic Structure Puzzle

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**a e u i**  
**y t w l**  
**t e e e**

# Atomic Structure Puzzle

Number of electrons in a neutral phosphorus atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Atomic number of copper  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of protons in $^{18}\text{O}^{2-}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Mass number of an atom with 12 protons, 12 electrons, and 12 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>
Number of electrons in $\text{S}^{2-}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of neutrons in an $^{56}\text{Fe}^{3+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of neutrons for $^{60}\text{Ni}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of protons in a neutral atom with 18 neutrons and 16 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>
Number of neutrons in $^{59}\text{Ni}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of neutrons in $^{68}_{30}\text{Zn}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of protons in an $^{27}\text{Al}^{3+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Mass number of a neutral atom with 17 electrons and 18 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>
Mass number of a $-3$ ion with 7 neutrons and 10 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of protons in a manganese atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of electrons in a $\text{Sr}^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Number of electrons in a neutral cobalt atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>
Atomic number of a $+4$ ion having 26 neutrons and 18 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Atomic number of $^{39}_{19}\text{K}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Mass number of an ion with 21 protons, 19 electrons, and 24 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>	Mass number of $^{28}_{14}\text{Si}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px;">Set 7</span>

# Atomic Structure Puzzle

**f o h c**

**o a e u**

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**r t o c**

**t o n t**

# Atomic Structure Puzzle

Number of neutrons in $^{105}_{46}\text{Pd}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Number of neutrons for $^{91}\text{Zr}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Atomic number of indium  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Mass number of an atom with 32 protons, 32 electrons, and 40 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>
Number of protons in $\text{Sn}^{2+}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Number of electrons in a neutral nickel atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Mass number of a $\text{Cl}^-$ ion with 20 neutrons and 18 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Mass number of $^{62}_{28}\text{Ni}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>
Atomic number of yttrium  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Number of electrons in a neutral atom with atomic number 25  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Number of protons in a $^{80}\text{Se}^{2-}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Number of protons in an element with atomic number 32  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>
Number of electrons in $^{56}\text{Fe}^{3+}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Number of electrons in a $\text{Br}^-$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Atomic number of a $-2$ ion having 18 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Mass number of a neutral atom with 22 neutrons and 20 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>
Number of protons in a $^{65}\text{Zn}^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Atomic number of $^{91}_{40}\text{Zr}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Number of protons in a neutral atom with 49 neutrons and 38 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>	Mass number of a $+1$ ion with 46 electrons and 62 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 8</span>

# Atomic Structure Puzzle

**i A o u**

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**f e e**

**d c . e**

# Atomic Structure Puzzle

Number of protons in $O^{2-}$  _____ Set 9	Number of neutrons in ${}^7Li$  _____ Set 9	Number of neutrons in ${}^{20}Ne$  _____ Set 9	Mass number of an atom with 5 protons, 5 electrons, and 6 neutrons  _____ Set 9
Atomic number of hydrogen  _____ Set 9	Number of electrons in a neutral ${}^{11}B$ atom  _____ Set 9	Number of electrons in a $Sr^{2+}$ ion  _____ Set 9	Atomic number of a +2 ion having 10 electrons  _____ Set 9
Number of neutrons in ${}^{31}_{15}P$  _____ Set 9	Number of protons in a neutral atom with 16 neutrons and 14 electrons  _____ Set 9	Number of protons in a carbon atom  _____ Set 9	Atomic number of ${}^{35}_{17}Cl$  _____ Set 9
Number of electrons in a neutral atom with atomic number 15  _____ Set 9	Number of electrons in a neutral nitrogen atom  _____ Set 9	Mass number of ${}^{19}_9F$  _____ Set 9	Mass number of a neutral atom with 20 neutrons and 18 electrons  _____ Set 9
Number of neutrons in a ${}^{41}K^+$ ion  _____ Set 9	Mass number of an ion with 8 protons, 10 electrons, and 10 neutrons  _____ Set 9	Mass number of a +2 ion with 34 neutrons and 26 electrons  _____ Set 9	Number of protons in an element with atomic number 13  _____ Set 9

# Atomic Structure Puzzle

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**h c m a**

# Atomic Structure Puzzle

<p>Number of protons in <math>\text{Sn}^{4+}</math></p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of neutrons in <math>^{70}\text{Ge}</math></p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of neutrons in <math>^{67}_{30}\text{Zn}</math></p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of neutrons in a <math>^{59}\text{Ni}^{2+}</math> ion</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>
<p>Atomic number of selenium</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of electrons in a <math>\text{Rb}^+</math> ion</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Mass number of an ion with 19 protons, 18 electrons, and 20 neutrons</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of electrons in a <math>^{32}\text{S}^{2-}</math> ion</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>
<p>Number of electrons in a neutral potassium atom</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of electrons in a neutral atom with atomic number 32</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Mass number of an atom with 21 protons, 21 electrons, and 24 neutrons</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of protons in an element with atomic number 33</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>
<p>Atomic number of a +2 ion with 13 neutrons and 10 electrons</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Atomic number of a -1 ion having 54 electrons</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of electrons in a <math>\text{Pb}^{2+}</math> ion</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of protons in an element with atomic number 35</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>
<p>Number of protons in a <math>^{111}\text{Cd}^{2+}</math> ion</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Atomic number of <math>^{68}_{30}\text{Zn}</math></p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Number of protons in a neutral atom with 72 neutrons and 51 electrons</p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>	<p>Mass number of <math>^{40}_{18}\text{Ar}</math></p> <p style="text-align: center;">_____</p> <p style="text-align: right; background-color: #cccccc;">Set 10</p>

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

Number of electrons in $P^{3-}$  _____ Set 11	Number of neutrons in $^{59}_{27}\text{Co}$  _____ Set 11	Mass number of an atom with 9 protons, 9 electrons, and 10 neutrons  _____ Set 11	Number of electrons in a $\text{Cu}^{2+}$ ion  _____ Set 11
Number of electrons in a neutral titanium atom  _____ Set 11	Number of protons in $S^{2-}$  _____ Set 11	Number of neutrons in $^{29}\text{Si}$  _____ Set 11	Number of protons in $^{40}\text{Ca}^{2+}$  _____ Set 11
Atomic number for neon  _____ Set 11	Atomic number of a +6 ion having 18 electrons  _____ Set 11	Number of protons in a neutral atom with 6 neutrons and 6 electrons  _____ Set 11	Number of electrons in a neutral bromine atom  _____ Set 11
Number of electrons in a neutral $^{19}\text{F}$ atom  _____ Set 11	Number of neutrons in a $^{26}\text{Mg}^{2+}$ ion  _____ Set 11	Mass number of a +1 ion with 4 neutrons and 2 electrons  _____ Set 11	Number of protons in an element with atomic number 23  _____ Set 11
Mass number of an atom with 5 protons, 5 electrons, and 6 neutrons  _____ Set 11	Atomic number of $^{56}_{26}\text{Fe}$  _____ Set 11	Mass number of a +2 ion with 12 protons, 10 electrons, and 13 neutrons  _____ Set 11	Mass number of $^{21}_{10}\text{Ne}$  _____ Set 11

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

Mass number of an atom with 14 electrons, 14 protons, and 16 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of protons in $^{59}\text{Co}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of electrons in a neutral potassium atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of electrons in $^{31}\text{P}^{3-}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>
Atomic number of iron  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of neutrons in $^{23}_{11}\text{Na}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of protons in a $\text{Ca}^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of electrons in an $^{27}\text{Al}^{3+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>
Atomic number of a +2 ion having 27 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of protons in a $^{59}\text{Ni}^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of electrons in a neutral titanium atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Mass number of a neutral atom with 18 neutrons and 16 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>
Number of protons in $^{31}\text{P}^{3-}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of protons in an element with atomic number 9  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of protons in $^{52}\text{Cr}^{3+}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Atomic number of $^{51}_{23}\text{V}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>
Mass number of a $-1$ ion with 18 neutrons and 18 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Mass number of an ion with 15 protons, 18 electrons, and 16 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Number of electrons in a neutral atom with atomic number 25  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>	Mass number of $^{33}_{16}\text{S}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 12</span>

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

Number of protons in a neutral atom with 75 neutrons and 54 electrons  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Mass number of an atom with 64 protons, 64 electrons, and 92 neutrons  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Atomic number of erbium  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of neutrons in $^{203}\text{Tl}$  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>
Atomic number of a +1 ion having 54 electrons  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of neutrons in $^{178}\text{Hf}$  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of electrons in a neutral atom with atomic number 59  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of electrons for $\text{Fr}^+$  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>
Number of protons in a neutral atom with 120 neutrons and 80 electrons  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of protons in an element with atomic number 77  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of electrons in a neutral gold atom  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of neutrons in $^{137}\text{Ba}^{2+}$  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>
Mass number of a -1 ion with 54 electrons and 74 neutrons  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of electrons in a $\text{Pb}^{4+}$ ion  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Mass number of $^{130}_{54}\text{Xe}$  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Mass number of a neutral atom with 87 neutrons and 62 electrons  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>
Atomic number of $^{183}_{74}\text{W}$  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of electrons in a neutral gadolinium atom  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Mass number of an ion with 80 protons, 78 electrons, and 120 neutrons  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>	Number of neutrons in $^{193}_{77}\text{Ir}$  _____ <div style="text-align: right; background-color: #cccccc; padding: 2px;">Set 13</div>

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

Number of neutrons in $^{73}_{32}\text{Ge}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of electrons in a neutral bismuth atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of protons in a neutral atom with 70 neutrons and 51 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Mass number of a neutral atom with 120 neutrons and 80 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>
Atomic number of silicon  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of electrons in a neutral cadmium atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of neutrons in a $^{88}\text{Sr}^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of neutrons in $^{59}\text{Co}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>
Number of electrons in a $\text{O}^{2-}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Mass number of a +4 ion with 109 neutrons and 70 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of protons in $^{31}\text{P}^{3-}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of protons in a $^{44}\text{Ca}^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>
Number of protons in an element with atomic number 46  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of electrons for $\text{S}^{2-}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Atomic number of $^{99}_{44}\text{Ru}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Number of electrons in a neutral atom with atomic number 39  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>
Atomic number of a -1 ion having 36 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Mass number of $^{93}_{41}\text{Nb}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Mass number of an atom with 36 protons, 36 electrons, and 46 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>	Mass number of an ion with 29 protons, 27 electrons, and 34 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 14</span>

# Atomic Structure Puzzle

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# Atomic Structure Puzzle

Number of electrons in a neutral tellurium atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Mass number of a $-2$ ion with 9 neutrons and 10 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Number of electrons for $P^{3-}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Number of electrons in a neutral ytterbium atom  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>
Number of neutrons in $^{58}\text{Fe}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Atomic number of mercury  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Number of electrons in a neutral atom with atomic number 40  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Number of protons in a $^{122}\text{Sn}^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>
Number of electrons in a $\text{N}^{3-}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Atomic number of a $-2$ ion having 58 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Number of neutrons in $^{87}_{38}\text{Sr}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Atomic number of $^{78}_{34}\text{Se}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>
Mass number of a neutral atom with 98 neutrons and 67 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Number of protons in an element with atomic number 46  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Number of protons in $^{32}\text{S}^{2-}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Mass number of an ion with 28 protons, 26 electrons, and 32 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>
Number of neutrons in a $^{44}\text{Ca}^{2+}$ ion  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Mass number of an atom with 42 electrons, 42 protons, and 55 neutrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Mass number of $^{48}_{22}\text{Ti}$  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>	Number of protons in a neutral atom with 72 neutrons and 51 electrons  _____ <span style="float: right; background-color: #cccccc; padding: 2px 5px;">Set 15</span>

# Atomic Structure Puzzle

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